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(71) Applicant (for all designated States except US): **THE TRUSTEES OF COLUMBIA UNIVERSITY IN THE CITY OF NEW YORK [US/US];** 116th Street and Broadway, New York, NY 10027 (US).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **ADAMS, David**

[US/US]; 90 Morningside Drive, Apt. 2L, New York, NY 10027 (US). **CHERNIAVSKAYA, Oksana [US/US];** 419 West 118th Street, Apt. 23A, New York, NY 10027 (US).

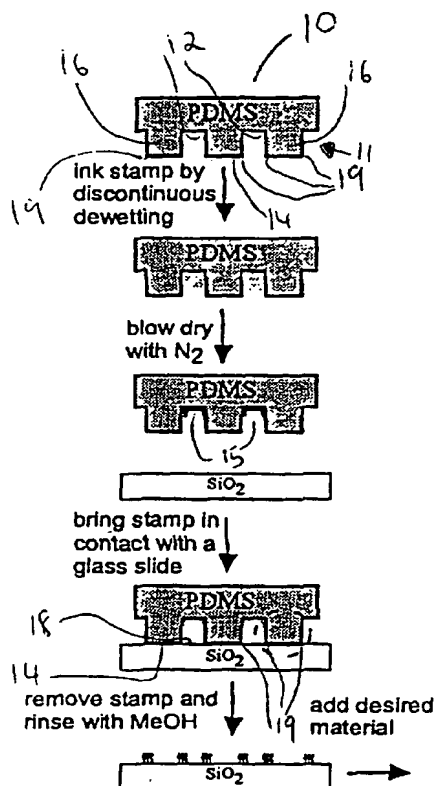
(74) Agents: **TANG, Henry et al.;** Baker Botts, LLP, 30 Rockefeller Plaza, New York, NY 10112 (US).

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(54) Title: **EDGE TRANSFER LITHOGRAPHY**



(57) Abstract: A method for applying a nanoscale resolution pattern of a molecular ink onto a surface of a substrate is disclosed. The method referred to as edge transfer lithography, comprises providing a stamp structure (10) having a surface (11) with at least one protruding feature (16). Each protruding feature (16) has a stamp surface of a respective predefined shape at a protruding end thereof. Each protruding feature and its stamp surface are bounded by at least one edge (19), which edge intersects the surface of the stamp structure to form a recess. A solution of the molecular ink and a solvent is applied of the surface of the stamp structure (10) are such that the solution dewets from the surface of the stamp structure (10) is then dried to evaporate the solvent.

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EDGE TRANSFER LITHOGRAPHY

SPECIFICATION

CROSS-REFERENCE TO RELATED APPLICATION:

This application claims priority from U.S. Provisional Application Ser.
5 No. 60/286,755, filed April 25, 2001..

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention is directed at a method for applying a molecular ink
onto a substrate surface. In particular, this invention is directed at a method for
10 applying self-assembled molecular monolayers onto a surface at high resolution.

2. Background Information

Lithographic methods have played a major role in the development of
modern microelectronics and are expected to be of central importance in the
developing fields of nanotechnology and molecular electronics. A current challenge
15 in the lithographic field is to control the lateral placement of molecules on surfaces
with a resolution, or line width, under 100 nm.

Two methods have been recently developed for direct writing and
patterning of surfaces with molecular nanostructures and self-assembled monolayers
(SAMs): dip-pen nanolithography (DPN), described, for example, in Piner, R. D.,
20 Zhu, J., Xu, F., Hong, S., and Mirkin, C. A., "Dip-Pen" Nanolithography," *Science*,
Vol. 283 (1999), pp. 661-63; and micro-contact printing (μ CP), described, for
example, in Xia, X., and Whitesides, G. M., "Soft lithography," *Angewandte Chemie
International Edition in English*, Vol. 37 (1998), pp. 550-74, and in U.S. Patent No.
6,060,121. μ CP utilizes a molded "stamp" to print "molecular inks" such as alkane
25 thiols on gold, as described, for example, in Jeong, N. L., Nuzzo, R. G., Xia, Y.,
Mrksich, M., and Whitesides, G. M., "Patterned Self-Assembled Monolayers formed

by *Microcontact Printing: Direct Selective Metalization by Chemical Vapor Deposition on Planar Substrates*," *Langmuir*, Vol. 11 (1995), pp. 3024-26, and siloxanes on silicon or silica surfaces, as described, for example, in Xia, Y., Mrksich, M., Kim, E., and Whitesides, G. M., "*Microcontact Printing of Octadecylsiloxane on the Surface of Silicon Dioxide and its Applications in Microfabrication*," *Journal of the American Chemical Society*, Vol. 117 (1995), pp. 9576-77. The term "stamp" as used herein refers to a structure having a surface with one or more features protruding from the surface, wherein each of the one or more protruding features has a stamp surface bounded by at least one edge, and each pair of adjacent protruding features defines an inner recess. The stamps can be prepared by using a silicon wafer having a desired pattern of grooves formed by photolithography as a mold. The resulting stamps have a structure which mirrors the topographic structure of the silicon wafer template. The term "molecular ink" as used herein refers to a substance that can be transferred from a stamp onto the surface of a substrate, such as glass, by contacting the stamp with the substrate surface. In the case of conventional μ CP, the stamp surfaces are wetted with the molecular ink so that the ink is transferred from the stamp surface to the surface of the substrate. The substance may be a compound or a mixture of compounds. μ CP may be used to make a wide variety of arrays of patterned SAMs on substrate surfaces. However, because the molecular ink is transferred from the stamp surfaces of the protruding features of the stamp, the line width which can be achieved with μ CP is typically only about 500 nm or higher, although sub-500 nm line widths have been reported by Delamarche, E., et al., "*Transport Mechanism of Alkanethiols during Microcontact Printing on Gold*," *Journal of Physical Chemistry B*, Vol. 102 (1998), pp. 3324-34. DPN employs an atomic force microscope (AFM) probe to deliver molecules from the probe surface to a substrate. It is believed that the delivery takes place via a meniscus of water between the probe and the substrate. DPN can achieve feature sizes on the order of 100 nm and is therefore superior to μ CP in this respect. However, DPN has the disadvantage of requiring expensive and sophisticated apparatus, and of serial "writing" of a pattern to be transferred to the surface of the substrate. Consequently,

the stamp structure surface by a common distance. Each protruding feature has a stamp surface of a respective predefined shape at a protruding end thereof. Each protruding feature and its stamp surface are bounded by at least one edge. Each edge intersecting the surface of the stamp structure to form a recess. A solution of the molecular ink and a solvent is applied to the surface of the stamp structure, including each stamp surface. The solution of the molecular ink and the material of the surface of the stamp structure are such that the solution dewets from the surface of the stamp structure, including each stamp surface, so as to accumulate in each recess. The surface of the stamp structure, including each protruding feature, are then dried to evaporate the solvent and leave the molecular ink in each recess with substantially no molecular ink on each stamp surface. Each stamp surface is then brought into contact with the surface of the substrate to transfer the molecular ink from each recess to the surface of the substrate along each edge of each protruding feature so as to apply the nanoscale resolution pattern of the molecular ink onto the surface of the substrate, wherein the molecular ink is covalently bound to the surface. A monolayer of the substance is then deposited from a solution containing the substance on the pattern of molecular ink on the surface of the substrate by using one of casting, spin-coating and dip-coating.

The method of the invention, which is referred herein as edge transfer lithography, or ETL, maintains the simplicity and broad applicability of μ CP and at the same time substantially reduces the feature sizes which are possible with the μ CP approach because molecular ink is transferred to the substrate surface along each edge of each protruding feature, rather than from each stamp surface. A resolution of up to 60 nm line width can be obtained with ETL. ETL is relatively simple to implement, in contrast to DPN and other nanometer patterning techniques such as E-beam lithography, AFM and STM (scanning tunneling microscopy) modification of surfaces, each of which requires expensive and sophisticated apparatus for their implementation. Accordingly, ETL provides a simple and widely applicable way of patterning the surface of a substrate with nanoscale features over relatively large areas. ETL may also be used to form high-order junctions through multiple applications of the ETL stamping process.

The “molecular ink” is preferably a substance containing a compound capable of forming SAMs on the substrate surface and which dewets from the stamp structure surface, including each stamp surface. The patterned SAMs render possible the guided assembly of molecular materials. Therefore, ETL may also be used to
5 create complex patterned nanoscale 3-dimensional structures, with applications in molecular and nanoscale electronics, chemosensing, surface science, catalysis, and the biological sciences.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1A is a schematic illustration of the ETL process of the present
10 invention.

Figure 1B is a schematic illustration of a possible mechanism for molecular transport of alkylsilanes from the edge of a stamp feature to the glass substrate via an adsorbed water meniscus.

Figure 1C shows a reverse scan direction LFM 3-D image of a 1.5 x
15 1.5 μm clover-shaped SAM of dimethyldodecylchlorosilane (DDCS) deposited onto a glass surface.

Figures 1D-E show idealized structures of a liquid crystal/SAM bilayer assembly formed in accordance with the present invention.

Figures 2A-2E show scanning probe microscopy images of self
20 assembled monolayers of DDCS patterned on glass using an elastomeric stamp.

Figures 2F-2H show images of liquid crystal/SAM assemblies obtained by overlaying DDCS self-assembled monolayers with a liquid crystal monolayer in accordance with the present invention.

Figure 2I shows images of self assembled monolayers of DDCS
25 patterned on glass using an elastomeric stamp with a line pattern in accordance with the present invention.

Figure 2J shows images of self assembled monolayers of titanium dioxide patterned on glass using an elastomeric stamp in accordance with the present invention.

Figures 2K and 2L show NC-AFM images of CdSe nanoparticles overlaying DDCCS self-assembled monolayers formed in accordance with the present invention.

Figures 3A-E shows topographic line scans obtained from NC-AFM
5 images.

DETAILED DESCRIPTION OF THE INVENTION

Referring to Figure 1A, stamp structure 10 used in the method of the invention comprises a surface 11 having protruding features 16 each having a respective stamp surface 14 of a predetermined shape bounded by one or more edges
10 19, each extending from the stamp surface 14 to a respective non-protruding portion of the stamp structure surface 11. Each protruding feature extends from the stamp structure surface 11 by a common distance. The intersection 19 of each edge 14 and the stamp structure surface 11 forms a recess 12.

In an exemplary embodiment of the present invention, each recess 12
15 has a scale on the order of a micron. In an advantageous embodiment of the invention, the stamp structure surface 11 is formed with an elastomeric material. In an especially advantageous embodiment, the stamp structure surface 11 is made from a polymeric material. An especially advantageous polymeric material is polydimethoxysilane (PDMS). The PDMS stamps may be molded using a silicon
20 wafer having a desired pattern of grooves formed by photolithography or other standard methods as the mold. After molding, the stamp is cleaned with chloroform to remove any low molecular weight polymer from the stamp. The protruding features 16 and their respective stamp surfaces 14 can be of any shape. For illustrative purposes, stamp surfaces 14 of each of the protruding features 16 are
25 shown in Figure 1A as having an essentially rectangular shape.

The solvent in which the molecular ink is dissolved is a polar solvent. In an advantageous embodiment, the polar solvent is a solvent that readily dewets from the PDMS surface and causes substantially no swelling of the PDMS surface. An alcohol such as ethanol is advantageous over a non-polar solvent such as hexane,
30 hexadecane and toluene, which does not dewet readily from the PDMS surface and which swells the PDMS stamp structure surface 11. The stamp structure surface 11,

including each stamp surface 14, is typically inked with a fresh solution of the molecular ink in a concentration ranging between 1% and 5% by volume in anhydrous ethanol or methanol, whereupon the solution of the molecular ink discontinuously dewets from the stamp structure surface 11, including the stamp surface 14 of each protruding feature 16, and leaves molecules of the solvent and of the molecular ink 15 accumulated in the recesses 12 of the stamp structure surface 11, as shown in Figure 1A. Discontinuous wetting and dewetting has been described in the context of filling arrays of PDMS microwells in Jackman, R. J., Duffy, D. C., Ostuni, E., Willmore, N. D., and Whitesides, G. M., *Analytical Chemistry*, Vol. 70 (1998), pp. 2280-87. After the molecular ink solution is applied to the stamp structure surface 11, the surface 11 is blown dry preferably using nitrogen gas so as to evaporate the solvent and leave only a small amount of molecular ink on the stamp surfaces 14 of the protruding features 16, while a large reservoir of molecular ink is left in the recesses 12 of the stamp structure 11. When the stamp surfaces 14 of the protruding features 16 of the stamp structure surface 11 are brought into contact with substrate surface 18, the molecular ink is delivered from the recesses 12 along the edges 19 of the protruding features 16 so as to produce on the surface 18 of the substrate large area patterns with nanoscale features.

The molecular ink is a compound capable of forming self-assembled monolayers and of reacting with hydroxyl groups, which may be hydroxyl groups on a substrate surface 18, and with water, to form a covalent bond with the oxygen atom. In one advantageous embodiment, the compound is an alkylsilane substituted with a polar group. The polar group can be, for example, a leaving group, where the term "leaving group" as used herein refers to a functional group that can be displaced by a nucleophilic group. In one particularly advantageous embodiment, the alkylsilane is selected from the group consisting of monochloro alkylsilanes and monoalkoxy alkylsilanes. Among monoalkoxy alkylsilanes, monomethoxy alkylsilanes are especially advantageous. In another exemplary embodiment, the molecular ink compound is a metallic compound that is soluble in a liquid alcohol or a mixture of water and a liquid alcohol. One such advantageous metallic compound is titanium dioxide, TiO_2 , more particularly titanium dioxide nanoparticles ranging from about 3

to about 7 nm in diameter. These nanoparticles may be dissolved in ethanol or in ethanol:water mixtures in which the volume percentage of water in the solvent varies from about 0.02% to about 3% by volume. The concentration of titanium dioxide varies from about 3×10^{-3} to about 9×10^{-4} mg/ml. For the above values of water
5 percentage and titanium dioxide concentration, patterns having narrow line widths of titanium dioxide ink transferred from the edges 19 of the protruding features of the stamp structure surface 11 are obtained. At higher levels of water percentage and titanium dioxide concentration, more titanium dioxide ink is left on the stamp surfaces 14 of the protruding features 16 of the stamp structure surface 11 so as to produce
10 more "filled-in" patterns on the substrate surface 18.

In a further exemplary embodiment of the invention, the molecular ink is capable of forming on the substrate surface 18 a monolayer onto which may be deposited a monolayer of a substance that is not covalently bound to the molecular ink. Preferably, the molecular ink is covalently bound to the surface groups on the
15 substrate surface 18, such as hydroxy groups. In an advantageous embodiment of the invention, the substance that is not covalently bound to the molecular ink may be a liquid crystal material. Perylene derivatives are especially suitable liquid crystal materials for this purpose. In another advantageous embodiment of the invention, the substance that is not covalently bound to the molecular ink may be a semiconductor.
20 Cadmium-selenium (CdSe) nanoparticles represent an especially suitable semiconductor material for this purpose.

The substrate of the invention may be a conductor, such as a metal, a non-conductor, or a semiconductor. In an advantageous embodiment, the substrate is made of glass, a polymeric material, or an inorganic material such as a ceramic. In an
25 especially advantageous embodiment, the substrate is a glass slide having a surface that is freshly cleaned with a strong oxidant, such as a concentrated solution of sulfuric acid and hydrogen peroxide in water. In another advantageous embodiment, the glass slide and the stamp surface or surfaces 14 are brought into contact immediately after the drying or blow-drying step described above. Referring to
30 Figure 1A, the glass slide surface 18 and the stamp surfaces 14 of the protruding features 16 of the stamp 10 are brought into contact to cause the transfer of the

molecular ink from the stamp structure surface 11 to the glass slide surface 18 along the edges 19 of the protruding features 16. In a further embodiment of the invention, the stamp structure 10 and glass slide are baked while in contact at 70 °C for at least one hour. After the stamp is removed from the glass slide, the glass slide may be
5 further baked at about 70 °C for at least one hour, preferably between one and 12 hours, to further bind the molecular ink to the glass surface 18.

The stamp structures 10 used in the invention have protruding feature sizes of about 1 to about 5 μm , where the term "feature size" as used herein denotes the width 11 of the stamp surface 14 of each protruding feature 16 in Figure 1A.
10 PDMS stamps with stamp surface sizes below about 2 μm are observed to have significantly reduced contact areas. The result appears to be due to solvent swelling which makes the stamp surfaces 14 of the protruding features 16 concave thereby reducing the contact area of the stamp surfaces 14 of the protruding features 16 with the substrate surface 18, so that virtually all the molecular ink is transferred from the
15 edges of the protruding features.

Using ETL, patterns having nanoscale resolution are formed primarily by the transfer of molecular ink along the edges 19 of the protruding features 16 of the stamp structure surface 11 rather than from the stamp surfaces 14. As a result, the pattern that is formed on the substrate surface reflects the outline of the pre-existing
20 patterns of the stamp surfaces 14. For example, lateral force microscopy (LFM) images, shown in Figures 2A and 2B, show that clover-shaped SAM structures are formed on a surface of a glass substrate when using a stamp structure 10 having stamp surfaces 14 in the form of an array of 1 μm clover-shaped wells. The molecular ink used to form the patterns shown in Figures 2A and 2B is dimethyldodecylchlorosilane
25 (DDCS). No external pressure is applied to the stamp structure 10 during transfer of the molecular ink to form the patterns of Figures 2A and 2B. LFM is a contact mode scanning probe method which is sensitive to chemical variations at the surface and has been used to image SAMs as described, for example, in Gauthier, S., Aime, J. P., Bouhacina, T., Attias, A. J., and Desbat, B., "*Study of Grafted Silane Monolayers on*
30 *Silica Surface with an Atomic Force Microscope*," *Langmuir*, Vol. 12 (1996), 5126-37. LFM data were collected using ThermoMicroscopes Explorer (probe part #1500-

stamp contacted the substrate surface 18. The measured height of the outline of the pattern is less than the 1.6 nm expected for a tightly packed SAM of dodecyl siloxane covalently bound to and projecting vertically from the surface of the substrate, which suggests a certain degree of tilt or disorder which is not uncommon in such structures.

5 In contrast, the average height in the region where the PDMS stamp surface 14 rested on or near the substrate during stamping measures only 0.2 nm above the bare glass surface. This very small value is only slightly above the noise level of the instrument and is believed to be due to partial silylation in this region, with molecules lying about parallel, rather than perpendicular, to the substrate surface. These data clearly
10 indicate that discontinuous dewetting of PDMS stamps with alkylsilane alcohol solutions leaves the alkylsilane available for delivery along the edges of the protruding features 16 of the stamp structure surface 11, with little, if any, alkylsilane on the stamp surfaces 14 of the protruding features. Similar data were obtained for TiO₂. Figure 3D shows a topographic line scan plot obtained for the pattern whose
15 image is shown in Figure 2J. The observed AFM height of slightly more than 5 nm indicates that a single layer of TiO₂ nanoparticles is present. Nearly all of the nanoparticles are delivered at the edges of the stamp surfaces 14, although a very small amount of deposition also occurs in the stamp surface contact region within the edges, unlike the case of alkylsilanes. A possible reason for this difference between
20 TiO₂ nanoparticles and alkylsilanes is that alkylsilane molecules on the surface of the protruding features 16 of the stamp structure 10 may degrade by exposure to water and may be absorbed into the PDMS stamp, while TiO₂ nanoparticles remain unaffected on the stamp surfaces 14 of the protruding features 16 of the stamp structure 10 and may be therefore transferred to the substrate.

25 Self-assembled monolayers can also be used for the guided assembly of molecular materials, as discussed, for example, in Huang, Z., Wang, P. C., MacDiarmid, A. G., Xia, Y., and Whitesides, G. M., *"Selective Deposition of Conducting Polymers on Hydroxyl-Terminated Surfaces with Printed Monolayers of Alkylsiloxanes as Templates,"* *Langmuir*, Vol. 13 (1997), 6480-85; Gupta, V. K., and
30 Abbott, N. L., *"Using Droplets of Nematic Liquid Crystals to Probe the Microscopic and Mesoscopic Structure of Organic Surfaces,"* *Langmuir*, Vol. 15 (1999), 7213-23;

Drawhorn, R. A., and Abbott, N. A., "Anchoring of Nematic Liquid Crystals on Self-Assembled Monolayers formed from Alkanethiols on Semitransparent Films of Gold," *Journal of Physical Chemistry*, Vol. 99 (1995), 16511-15; Aizenberg, J., Black, A. J., and Whitesides, G. M., "Control of Crystal Nucleation by Self-Assembled
5 *Monolayers*," *Nature*, Vol. 398 (1999), 495-97; and Liu, J., et al., "Controlled
*Deposition of Individual Single-Walled Carbon Nanotubes on Chemically
Functionalized Templates*," *Chemical Physics Letters*, Vol. 303 (1999), 125-29. We
have found that deposition of monolayers of certain substances onto the patterned
surfaces results in complex patterned layered systems where molecules of the
10 substance selectively organize onto the nanoscale SAM regions. Preferred substances
are nanoparticles of metallic compounds, such as CdSe capped with trioctylphosphine
(TOPO), preferably about 3 – 6 nm in diameter, and liquid crystal materials, such as
perylene, and in particular N,N'-bis[3-[2-[2-(1-
butoxy)ethoxy]ethoxy]propyl]perylene-3,4,9,10-tetracarboxyldiimide (PPEEB). The
15 combined liquid crystal/SAM assemblies were imaged by NC-AFM. Figures 2F and
2G show NC-AFM images of patterns formed by depositing liquid crystal perylenes
from dilute solutions (< 1 mM) onto various SAM patterned surfaces. The liquid
crystal/SAM assembly of Figure 2F is prepared by casting a dilute solution of liquid
crystal perylene onto a substrate surface patterned by ETL with DDCS. The liquid
20 crystal solution is prepared as described by Cormier, R.A., and Gregg, B.A., "Self-
Organization in Thin Films of Liquid Crystalline Perylene Diimides," *Journal of
Physical Chemistry B*, Vol. 101 (1997), pp. 11004-06. "Casting" as used herein refers
to placing several drops of a dilute solution in a solvent of the substance to be
deposited on the patterned substrate which is then placed in an atmosphere saturated
25 with the solvent, thereby ensuring slow evaporation of the solvent from the substrate.
The solvent used is typically tetrahydrofuran (THF) for perylene derivatives and
chloroform for CdSe particles. Preferred casting concentrations are about 0.01 mg/ml
for a perylene such as PPEEB in THF, and about 10^{-7} M for CdSe in chloroform. The
liquid crystal/SAM assembly of Figure 2G is prepared by spin-coating a liquid crystal
30 solution of perylene onto a surface patterned by ETL. "Spin-coating" as used herein
refers to placing at least one drop of a dilute solution in a solvent of the substance to

ETL can be used to form a molecular ink pattern containing arrays of four-way or higher order junctions by multiple applications of one or more stamps. Figure 2H shows a NC-AFM image of a pattern containing four-way junctions created by applying on a glass substrate stamp surfaces 14 in the form of 1 μm wide lines followed by stamp surfaces 14 in the form of 1.5 μm diameter circles. The order in which the stamp surfaces are applied is not critical. The white arrow points at a four-way junction. A DDCS concentration of about 1% was used to pattern the substrate, resulting in line widths of only 50 – 60 nm thickness. A very dilute solution of liquid crystal perylene diimide was spin-coated onto the substrate to enhance the height of the features shown in Figure 2H. Therefore, ETL can be used to form a pattern on a substrate surface by performing the steps of the method of the invention described above using a first material and a first molecular ink dissolved in a first solvent, and repeating these steps a multiple number of times, each time using a material, molecular ink, and solvent which may be the same as or different from a previously used material, molecular ink, and solvent, respectively. Piezo-driven nanopositioning and stamping would allow for precision multi-ink printing similar to a printing press. The ability to stamp multiple times allows for the possibility of applying different inks on the same surface to create multi-functional patterns, similar to multi-pen DPN.

The line widths obtained with ETL depend on humidity. Narrower lines were observed when ETL was performed under less humid conditions. Relative humidity likely affects the transport rate of alkylsilane to the surface, as in DPN. Water has also been shown to be essential for the reactivity of chloro- and methoxy-silanes with silica to form stable SAMs, as discussed, for example, in Angst, D. L., and Simmons, G. W., "Moisture Absorption Characteristics of Organosiloxane Self-Assembled Monolayers," *Langmuir*, Vol. 7 (1991), 2236-42. The size of the water meniscus is also critically dependent on the humidity, as disclosed in Piner, R. D., and Mirkin, C. A., "Effect of Water on Lateral Force Microscopy in Air," *Langmuir*, Vol. 13 (1997), 6864-68. The ink concentration, stamp substrate contact time, pressure applied to the stamp, SAM annealing procedure, nature of the silane reagent, and the substrate condition/topography all affect the line widths and SAM structure.

Example 2. Application of dimethyldodecylchlorosilane
(DDCS) to a glass surface

The experiment is performed in the same way as in Example 1 but using DDCS (90% pure used as purchased from Fluka) instead of DOMS.

5

It should be understood that various changes and modifications to the preferred embodiments described herein will be apparent to those skilled in the art. Such changes and modifications can be made without departing from the spirit and
10 scope of this invention, the scope being defined by the appended claims.

5. The method of Claim 1, wherein the solvent is a polar solvent.
6. The method of Claim 5, wherein the polar solvent causes substantially no swelling of the surface of the stamp structure and each protruding feature thereof.
7. The method of Claim 1, wherein the molecular ink is a compound
5 capable of forming self-assembled monolayers on the substrate surface and capable of reacting with hydroxy groups and with water.
8. The method of Claim 7, wherein the compound capable of forming self-assembled monolayers is an alkylsilane with a polar group.
9. The method of Claim 8, wherein the alkylsilane is selected from the
10 group consisting of monochloro alkylsilanes and monoalkoxy alkylsilanes.
10. The method of Claim 7, wherein the compound capable of forming self-assembled monolayers is a metallic compound soluble in one of a liquid alcohol and a mixture of water and a liquid alcohol.
11. The method of Claim 10, wherein the metallic compound is titanium
15 dioxide.
12. The method of Claim 11, wherein the titanium dioxide is in the form of nanoparticles ranging from about 3 to about 7 nm in diameter.
13. The method of Claim 1, wherein step (c) comprises blow drying the surface of the stamp structure, including each protruding feature thereof, to evaporate
20 the solvent.
14. The method of Claim 1, wherein the substrate comprises a material selected from the group consisting of glass, a polymeric material, and a ceramic.
15. The method of Claim 1, wherein no external pressure is applied to the stamp structure or the substrate in step (d).

16. A method for applying a nanoscale resolution pattern of a molecular ink onto a surface of a substrate and depositing a monolayer of a substance on the pattern of the molecular ink on the surface of the substrate comprising:

- 5 a) providing a stamp structure having a surface with at least one protruding feature, each protruding feature extending from the surface of the stamp structure by a common distance, each protruding feature having a stamp surface of a respective predefined shape at a protruding end thereof, each protruding feature and its stamp surface being bounded by at least one edge, each edge intersecting the surface of the stamp structure to form a recess;
- 10 b) applying a solution of the molecular ink and a solvent to the surface of the stamp structure, including each stamp surface, wherein the solution and the material of the surface of the stamp structure are such that the solution dewets from the surface of the stamp structure, including each stamp surface, so as to accumulate in each recess;
- 15 c) drying the surface of the stamp structure, including each stamp surface, to evaporate the solvent and leave the molecular ink in each recess while leaving substantially no molecular ink on each stamp surface;
- 20 d) bringing each stamp surface in contact with the surface of the substrate to transfer the molecular ink from each recess to the surface of the substrate along each edge of each protruding feature so as to form a nanoscale resolution pattern of the molecular ink on the surface of the substrate, wherein the molecular ink is covalently bound to the surface of the substrate; and
- 25 e) depositing a monolayer of the substance on the pattern of the molecular ink on the surface of the substrate from a solution containing the substance using one of casting, spin-coating and dip-coating.

17. The method of Claim 16, wherein the material of the surface of the stamp structure, including each protruding feature thereof, is an elastomeric material.

- 5 a) providing a stamp structuring having a surface with at least one protruding feature, each protruding feature extending from the surface of the stamp structure by a common distance, each protruding feature having a stamp surface of a respective predefined shape at a protruding end thereof, each protruding feature and its stamp surface being bounded by at least one edge, each edge intersecting the surface of the stamp structure to form a recess;
- 10 b) applying a solution of the molecular ink and a solvent to the surface of the stamp structure, including each stamp surface, wherein the solution and the material of the surface of the stamp structure are such that the solution dewets from the surface of the stamp structure, including each stamp surface, so as to accumulate in each recess;
- 15 c) drying the surface of the stamp structure, including each stamp surface, to evaporate the solvent and leave the molecular ink in each recess while leaving substantially no molecular ink on each stamp surface; and
- 20 d) bringing each stamp surface in contact with the surface of the substrate to transfer the molecular ink from each recess to the surface of the substrate along each edge of each protruding feature so as to apply a nanoscale resolution pattern of the molecular ink on the surface of the substrate.

29. The method of Claim 28, wherein the material of the surface of the stamp structure, including each protruding feature thereof, is an elastomeric material.

25 30. The method of Claim 28, wherein the material of the surface of the stamp structure, including each protruding feature thereof, is a polymeric material.

31. The method of Claim 30, wherein the polymeric material is polydimethoxysilane.

32. The method of Claim 28, wherein the solvent is a polar solvent.

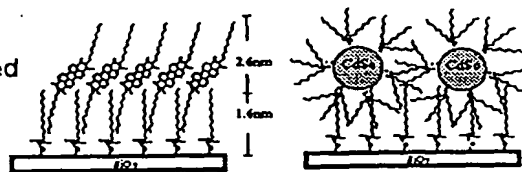
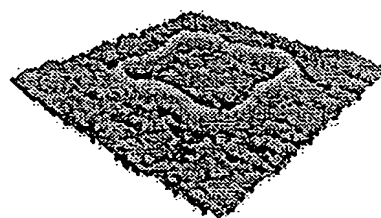
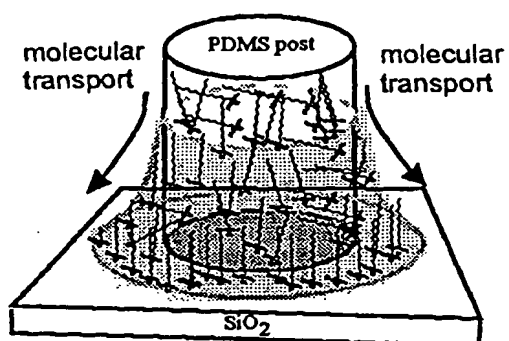
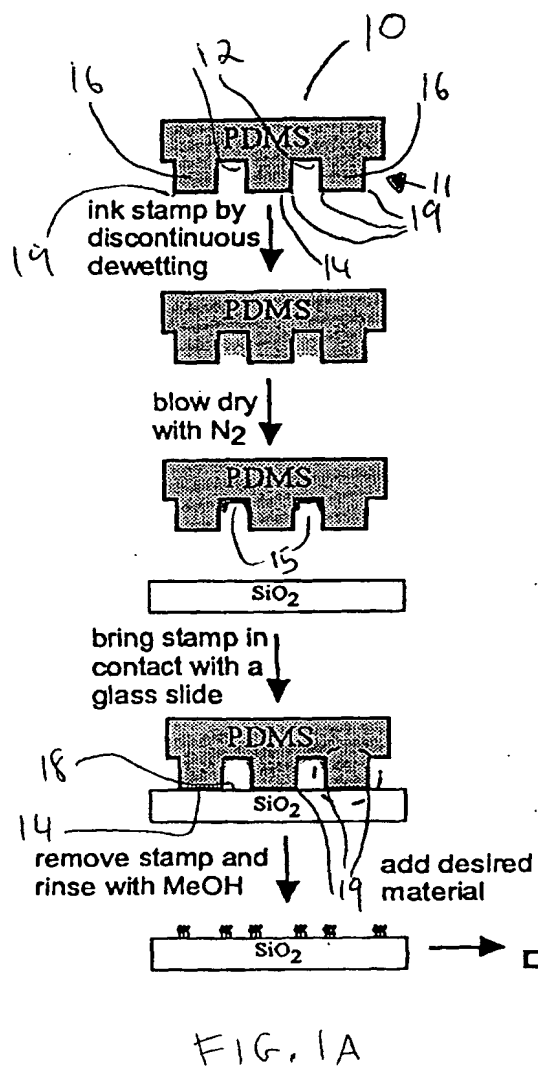


FIG. 2A

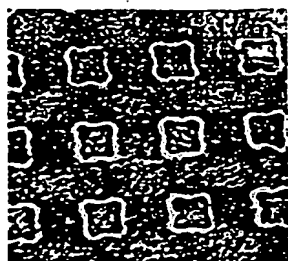


FIG. 2B

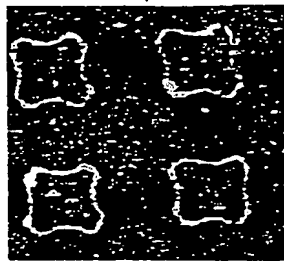


FIG. 2C

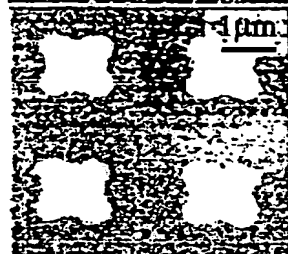


FIG. 2D

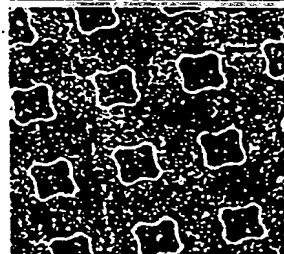


FIG. 2E

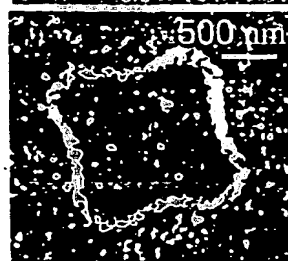


FIG. 2F

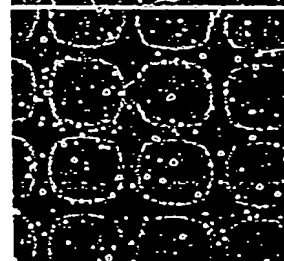


FIG. 2G

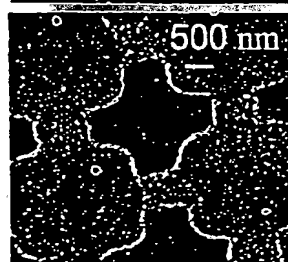


FIG. 2H



FIG. 2I

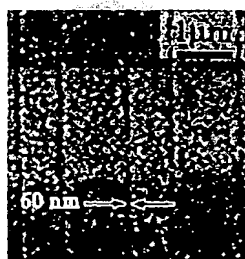


FIG. 2J

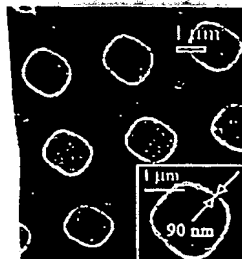
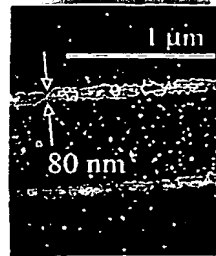


FIG. 2K



FIG. 2L



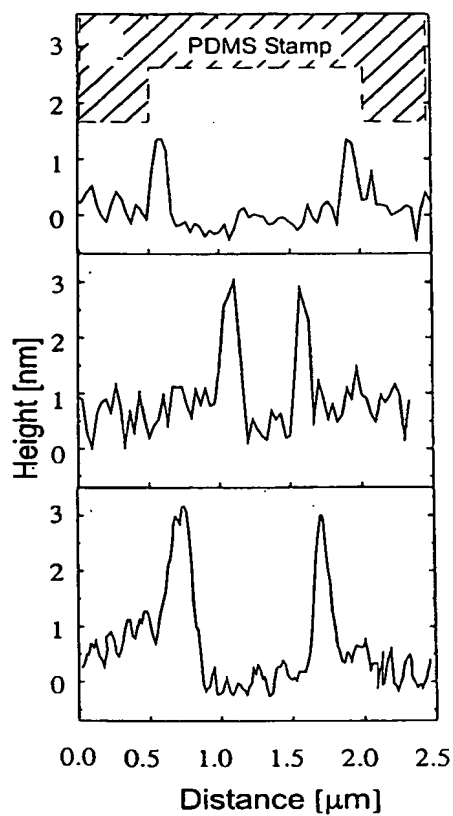


FIG. 3A

FIG. 3B

FIG. 3C

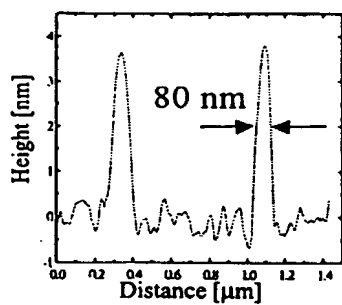


FIG. 3D

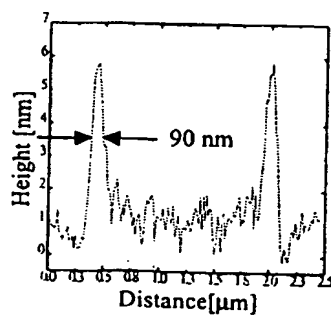


FIG. 3E

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US02/13164

A. CLASSIFICATION OF SUBJECT MATTER

IPC(7) :B41N 6/00

US CL :101/401.1, 483, 491, 493, 327, 150, 170

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 101/401.1, 483, 491, 493, 327, 150, 170

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
NONEElectronic data base consulted during the international search (name of data base and, where practicable, search terms used)
NONE

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5,669,303 A (MARACAS et al) 23 September 1997 (23.09.1997), see entire document.	1-42
A	US 6,119,596 A (FLETCHER et al) 19 September 2000 (19.09.2000), see entire document.	1-42
A	US 6,089,853 A (BIEBUYCK et al) 18 July 2000 (18.07.2000), see entire document.	1-42
A	US 5,937,758 A (MARACAS et al) 17 August 1999 (17.08.1999), see entire document.	1-42



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
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"L" document which may throw doubt on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"A" document member of the same patent family
"O" document referring to an oral disclosure, use, exhibition or other means	
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

03 JULY 2002

Date of mailing of the international search report

05 SEP 2002

Name and mailing address of the ISA/US
Commissioner of Patents and Trademarks
Box PCT
Washington, D.C. 20231

Facsimile No. (703)308-3230

Authorized officer

EUGENE H. EICKHOLT

Telephone No. (703)308-0125